

Lab 1, Introduction to Chain Shape and Physical Properties
Lab report information

Make sure that your report addresses the following questions, based on the first lab experiments.

Experiment A: Shape of a polymer chain

Conduct the data analysis as directed in the lab instructions. Here are some guidelines / reminders:

- The average of the squared end-to-end distance is calculated as $\langle r^2 \rangle = \frac{1}{N} \sum_i (r_i)^2$, where r_i is the distance that you measured in experiment i (so i ranges from 1 to 10 or 20). Repeat the calculation for each different kind of distance (end-to-end and maximum extent) and for each string.
- You may need to figure out the distribution of sizes “by hand”, such as by picking a width that seems to group multiple r values together (but not too many). The axes on a plot of the distribution will be the size of r (x-axis) and the scaled fraction of attempts that were within that size r (y-axis, see lab instructions for the scaling factors). An example of a distribution is the number of times that a pair of dice add up to a value between 2 and 12, after many rolls. Results for 36 rolls could be (if there was a perfect mix of rolls, which would not be likely to happen, in practice).

sum	number of rolls
2	1
3	2
4	3
5	4
6	5
7	6
8	5
9	4
10	3
11	2
12	1

- The combination of parameters $n\ell^2$ equals the value of $\langle r^2 \rangle = \frac{1}{N} \sum_i (r_i)^2$ that you determined from your experiment. This relationship emerges from the statistical mechanics of polymers. The fully extended length of the string corresponds to $n\ell$. n corresponds to the number of independent bond units along the backbone, and ℓ equals the length of each unit. Note that you can manipulate your results to estimate the values of n and ℓ that are consistent with each string. How do the results differ among the different strings?
- How do the calculation results compare between the end-to-end distance compared to the “maximum extent” distance?

Experiment B: Thermal effects on polymer properties

Answer the questions asked in the lab instructions:

- Rank the “bounciness” of the samples you tested, including temperature effects (see instructions).
 - How would you rank the “bounciness” of these samples?
 - Which bounces the most? Which bounces the least?
 - How do changes in temperature affect the bouncing?
- Rank the effects of heating on mechanical properties (see instructions).
 - What happens (if anything) when the sample is immersed in the boiling water? Describe.
 - How does the modulus (see below) compare for the same material under room temperature and 100°C conditions?
 - How does the modulus compare for these different samples and/or different polymers?

Experiment C: Sorption of liquid water

Answer the questions asked in the lab instructions:

- What did you observe as you added water to the diaper material?
- How much water could the diaper absorb? If you recorded the mass of the dry diaper material, how does the amount of absorbed water compare?
- Plot the expansion of the diaper material (longest length or approximate volume, depending on what you can calculate from your observations) vs. the mass of the added water.
- Describe your observations as you added salt to the hair gel. Compare the effect of adding salt to hair gel and to the polymer from the diaper.